MOLECULAR ASYMMETRY IN THE COORDINATION OF OLEFINS TO TRANSITION METALS: PRELIMINARY STUDIES OF CIS-DICHLORO -OLEFIN-AMINE-PLATINUM(II) COMPLEXES.

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In previous papers (1,2) we have described the preparation and properties of diastereoisemeric complexes of the general type: trans dichloro-olefin-a-phenethylamine-platinum(II)](i), where the sites of optical isometrism are the asymmetric (R or S) amine and an elefin (of symmetry lower than 2 mm; i.e. propylene, styrene, trans-2-butene) complexed to the metal atom.

No stereospecific effects of induction of configuration could be detected in solution. For instance, starting both from the (+)-trans-2-butene-(R)-amine and the (-)-trans-2-butene-(R)-amine complex the same equilibrium value of optical rotation was obtained, which corresponded in both cases to a 1:1 ratio of the two diastereoisomers. In this paper we wish to report some preliminary results concerning the corresponding cis complexes of

the general type : cis-[dichloro-clefin-a-phenethylamine-platinum(II)](ii).

To an aqueous solution of potassium tetrachleroplatinate (II) two moles of $(-)(S)\alpha$ -phenethylamine (with an optical purity of 97%) were added (3).

Crude (-) [cis dichlero-(S)- α -phenethylamine-platinum(II)], [Pt<C₈H₁₄H,Cl><C₈H₁₄H,Cl>], was recovered after standing one day. After boiling this product in 1 N,HCl, a yellow solution containing [C₈H₁₂H] [PtCl₃C₈H₁₄H] was obtained, from which, by shaking with ethylene at 50 psi green-yellow pale needles of (-) cis [dichlero-ethylene-(S)- α -phenethylene-platinum(II)] separated.

Recrystallisation from toluene gave a product having mp. 164°C , [a] ^{25}D - 54.5° (c 1.3, Acetone). Anal : Calcdt for $\text{C}_{10}\text{H}_{15}\text{MCl}_2\text{Pt}$; C 28.9; H 3.64; N.3.38. Found : C 28.71; H 3.84; N 3.69.

The IR spectrum of this complex is very similar to that of the trans-isomer (mp. 71°, [a] $^{25}_{D}$ - 33° (c 1.35, Acetone). From this compound we obtained the other olefinic complexes through exchange in solution with the corresponding olefin. As already observed in the preparation of the corresponding trans-complexes, the exchange yelds crude mixtures containing the two possible diastereoisomers, (+) and (-) cis-[dichloro-olefin-(S)a-phenethylamine-platinum (II)]. The resolution of the diastereoisomers of the pro-

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pylene, trans-2-butene and styrene complexes was obtained through crystallization from several different selvent mixtures.

The analyses of these compounds agreed with the assigned formula (ii).

X-Ray and Molecular Weight data of the (-) cis[trans- C_4H_8 , $C_8H_{11}^{\dagger}\Pi$, Cl_2Pt] are: a = 12.85, b = 8.88, c = 6.65 Å; γ = 102°38'; d 2.00 g/cm³; space group $P2_1$. The molecular weight of the independent unit is 446; mol.wt. (443, calcd.

The X-Ray structure analysis confirms the cis-coordination(4)
The complex with cis-2-butene was not capable of resolution,
because the two asymmetric carbon-atoms of the bonded olefin
have enantiomorphous configurations, and thus constitute a
meso-systems.

The optical rotation of the complexes are reported in Table I. In the trans series (1) the pure diastereoisomers of the complexes with trans-2-butene have values of molecular rotation equidistant from the value of molecular rotation of the corresponding ethylene complex. The activity of both diastereoisomers attain this value by epimerisation; this indicates that the solution at equilibrium containes a 1:1 ratio of the two diastereoisomers.

The diastereoisomeric complexes of the cis-series (ii), on the contrary, exhibit, equilibrium values of rotation signi-

	Table	I:	Data	of	optical	activity	a of			
cis-dicklero-clefin-(S)-a-phenethylamine-platinum(II)										

Olefin	Diastere [a] D	oisomers [X] _D	[a] _D b	[M] D b
ethylene			-54.5	226
cis-2-butene			-59.6	-264
(+)-propylene	+2.5	+10.7	-55.3	-238
(-)-propylene	-113	-485	-55.3	-238
(-)-trans-2-butene	-184	-815	-79.0	-350
(-)-styrene	-197	-9 67	-62.4	-306

a) In acctone at 25° (c 1.2 - 1.4)

ficantly different from that of the corresponding ethylene complex.

The effect is particularly noticeable in the case of the trans-butene complex (ii), where the molecular activity at equilibrium is nearly 50% higher than the activity of the ethylene complex.

We take these observations as an indication of a possible induction of asymmetry in the case of cis-complexes, in solution, due probably to a different steric interaction of the R or S complexed olefin in respect to the cis coordinated amine.

Further studies on this point are in progress.

References

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b) Values at equilibrium (after isomerisation) and values not-resoluble complexes.